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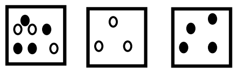

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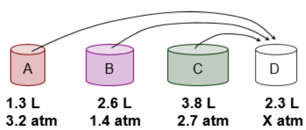

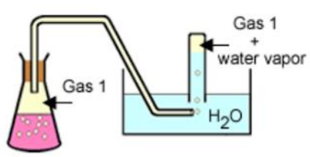

Hydrogen gas is collected over water at 22°C. Find the pressure of the dry gas if the atmospheric pressure is 708 mmHg.

A gas is collected over water at a temp of 35°C while the barometric pressure is 0.976 atm. What is the partial pressure of the dry gas?

N-33

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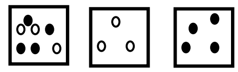

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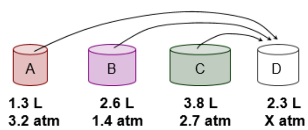

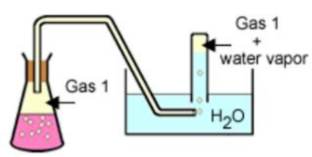

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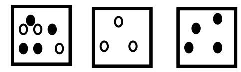

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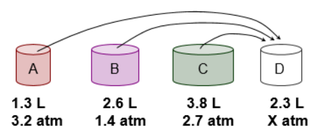

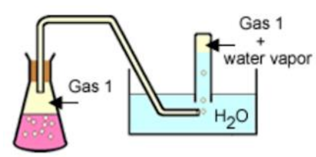

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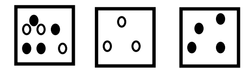

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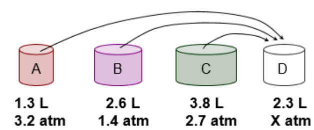

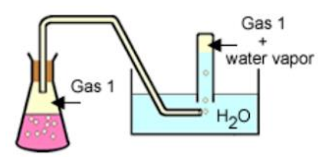

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